Electrochemical Valorisation of Cyrene Towards Fine Chemical Building Blocks





Ο

H. Kallweit, Erlangen/DE,^{1,2} P. Nikolaienko, Erlangen/DE,¹ K. J. J. Mayrhofer, Erlangen/DE^{1,2} 1) Forschungszentrum Jülich GmbH, Helmholtz Institute Erlangen-Nürnberg for Renewable Energy (IET-2), Cauerstr.1, 91058 Erlangen, Germany 2) Department of Chemical and Biological Engineering, Friedrich-Alexander-Universität Erlangen-Nürnberg, Egerlandstr. 3, 91058 Erlangen, Germany

| Chemical Industry | | | | |
|-------------------|-----------------------|----------|--|--|
| | STARTING MATERIALS | platform | traditional thermochemical transformations | |





- derived from cellulose waste
- industrially scalable production established



• bio-privileged molecule \rightarrow potential as platform chemical



- preparation: HT-CVD from methane and trimethyl borane
- parameter: boron doping, crystal size, sp² ratio, support material
- chemically inert
- low adsorption

synthesized by With at FAU

Voltmassogram



electrolyte: 0.1 M LiClO₄ in MeCN/H₂O = 1/1, 0 - 100 mM cyrene, WE: BDD on Si, A = 1cm², CE: Pt on Ti, RE: Ag/AgCl

Conclusion and Outlook

- Cyrene[™] can be oxidized by electrochemical means
- chiral lactone as platform chemical for further applications
 - \rightarrow derivatization of the starting material by aldol condensation to screen scope of electrochemical BVO
 - \rightarrow investigate other possible reactivities
- SFC-RTMS set-up enables high-throughput screening
 - \rightarrow parameter and material optimization via Design of Experiments or Bayesian Optimization
- insights from SFC-RTMS only semi-quantitative
 - \rightarrow transfer to larger scale flow cell
 - \rightarrow parallelization and automation to increase reproducibility and accelerate research
 - \rightarrow quantification of conversion, space-time yield, and Faradaic efficiency by coupling to HPLC



